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Synthesis and Biological Activity of some new Pyridazine Derivatives

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Summary. Condensation of β -aroyl- α -[1,3-diphenyl-5(4*H*)-oxo-pyrazol-4-yl] propionic acid with hydrazine hydrate affords 4,5-dihydro-3(2*H*)-pyridazinone (2). Reaction of 2 with POCl₃ and P₂S₅ gives a dichloro derivative (7) and a dithione (4). The behavior of the dichloro and dithione derivatives toward various reagents was studied. The *in vitro* antibacterial screening reveals moderate activities against certain bacteria.

Keywords. Pyridazinone; Pyridazinethione; Chloropyridiazine.

Synthese und biologische Aktivität einiger neuer Pyridazinderivate

Zusammenfassung. Kondensation von β -Aroyl- α -[1,3-diphenyl-5(4*H*)-oxo-pyrazol-4-yl]-propionsäure mit Hydrazinhydrat ergibt 4,5-Dihydro-3(2*H*)-pyridazinon (2). Reaktion von 2 mit POCl₃ und P₂S₅ liefert ein Dichlorderivat (7) und ein Dithion (4). Das Verhalten dieser beiden Verbindungen gegenüber verschiedenen Reagentien wurde untersucht. Antibakterielles screening (*in vitro*) ergab mäßige Aktivität gegenüber verschiedenen Bakterienstämmen.

Introduction

A recent publication dealing with the synthesis and antibacterial screening of pyridazinones revealed that some of these compounds exhibited activities against *gram*-positive and *gram*-negative bacteria [1-4]. In continuation of our studies [5, 6] on the synthesis of new pyridazinone compounds, we now report the preparation of a new series of pyridazinones to screen their antibacterial activities. The syntheses of various compounds (1-14) are outlined in Scheme 1.





Results and Discussion

The reaction of 3-(*p*-ethoxybenzoyl) acrylic acid [7] with 1,3-diphenyl-2-pyrazolin-5-one [8] in dry benzene gave 4,5-dihydro- α -[2-(4-ethoxyphenyl)-2-oxoethyl]-1,3diphenyl-5-oxo-1 *H*-pyrazole-4-acetic acid (1). The structure of the acid 1 was derived from its infrared spectrum # which showed v(C=O) (acid) at 1710, v(C=O) at 1665, and v(C=N) at 1595. Reaction of 1 with hydrazine hydrate in boiling ethanol yielded 6-(*p*-ethoxyphenyl)-4-(5-oxo-2-pyrazolin-4-yl)-4,5-dihydro-3(2*H*)-pyridazinone (2). The IR spectrum of 2 showed v(C=O) at 1660, v(C=N) at 1600, and v(NH) at 3430.

Compound 2 was reacted with *Lawsson* reagent in dry xylene. In an attempt to synthesize thienopyridazine, 3 instead dithioxo derivative 4 was obtained. Compound 4 evidently exists in a mercapto-thioxo equilibrium.

The IR spectrum exhibited bands for v(N-C=S) at 1470, v(C=S) at 1385, v(C=N) at 1600, and a band at 3420 (v(SH)). The ¹H NMR spectrum (*DMSO*-d₆) of **4** showed signals at $\delta = 7.7-7.1$ (15H, m, Ar-H + pyridazine proton), 3.43 (2H, q, CH₂ of ethyl), 3.30 (1H, s, pyrazolethione proton), and 1.34 (3H, t, CH₃ of ethyl) ppm.

The reaction of phenylmethanethiol with 7 gave a product identical with compound 5 obtained from the reaction of benzyl chloride with 4. The IR spectrum of 5 showed v(C=N) at 1605 and was devoid of v(C=S). Compound 4 was easily oxidized to the cyclic disulfide 6 by iodine solution; the IR spectrum of 6 showed v(C=N) at 1600. Treatment of 2 with POCl₃ gave the dichloro derivative 7.

The IR spectrum of 7 was devoid of v(C=O) and showed v(C=N) at 1605. The ¹H NMR spectrum (*DMSO*-d₆) of 7 showed signals at $\delta = 7.7-6.8$ (14H, m, Ar–H), 3.9–2.7 (5H, m, CH₂CH + CH₂ of ethyl), and 1.32 (3H, t, CH₃ of ethyl) ppm.

The dichloro derivative 7 reacted with sodium methoxide to give the dimethoxy derivative 8. The IR spectrum showed v(C=N) at 1610. Reaction of 7 with morpholine in dry toluene resulted in dechloroamination, affording the dimorpholino derivative 9 which showed v(C=N) at 1605. When compound 7 was allowed to react with hydrazine hydrate in ethanol, the dihydrazino derivative 10 was obtained instead of pyridazinopyridazine 11. The IR spectrum showed v(C=N) at 1600 and v(NH) at 3170.

Condensation of the hydrazino derivative 10 with aromatic aldehydes (ochlorobenzaldehyde, p-anisaldehyde, m-chlorobenzaldehyde, and benzaldehyde) gave the bis(hydrazone) derivatives 12a-d which showed v(C=N) at 1605–1600 and v(NH) at 3420–3400. The reaction of the dihydrazino derivative 10 with nitrous acid gave compound 14, presumably via azido derivative 13. The IR spectrum of 14 showed v(C=N) at 1590. An alternative route for the preparation of compound 14 was the reaction of the dichloro derivative 7 with sodium azide. The similarity of these compounds was identified by IR spectroscopy and mixed melting point analysis with the sample prepared before.

Screening for antibacterial activity

Compounds 2, 4, 5, 8, and 14 were tested for *in vitro* antibacterial activity using the method described by *Heatly* [9]. The medium for screening was composed of (g 11000 ml) "Lab-lemco" beef extract, 1.0; yeast extract (Oxoid 120), 20; peptone (Oxoid L37), 5.0; sodium chloride, 2.0; and agar, 15.0 (*pH* 7.0). Cylinders of known volume (0.1 ml) were placed on the solid medium seeded with a gram-positive and gram-negative test organism. A known constant volume (0.05 ml) of compounds 2, 4, 5, 8, and 14 dissolved in SDS was introduced into each cylinder, allowed to diffuse through the agar at room temperature for 1 h, and finally incubated at 37 °C for about 18–20 h. The zone of inhibition of bacterial growth around the disc was

Compound	Staphylococcus aureus	Echerichia coli	Pseudomona s auregonosa	Bacillus subtilis	Proteus
2	_	++		++	+++
4	~	+++	~	+	+ + +
5		++	+	+ + +	+
8		+	-	+	_
14	++	++	-	+++	+

Table 1. In vitro antibacterial activities of some of the prepared compounds^a

^a (-): No antibacterial activity of the zone equal to 1 cm; (+): mild activity with the diameter of the zone equal to 1 cm; (++): moderate activity with the diameter of the zone equal to 1.6 cm; (+++): good activity with the diameter of the zone equal to 2.1 cm

observed. The screening results given in Table 1 indicate that all the compounds exhibit antibacterial activities against various bacteria. Almost all compounds showed more inhibition against *E. coli*, *B. subtilis*, and *Proteus*, whereas inhibition towards *Staphylococcus aureus* and *Pseudomonas auregenosa* was less.

Experimental

All melting points are uncorrected. The IR spectra (KBr discs) were recorded on a Unicam SP 1200 spectrophotometer. The ¹H NMR spectra were obtained on a JEOL FX-100 fourier transform instrument using TMS as internal standard. Physical data are summarized in Table 2.

β -(4-Ethoxybenzoyl)- α -(1,3-diphenyl-5(4H)-oxo-pyrazolin-4-yl) propionic acid (1)

To a solution of 3-(p-ethyoxybenzoyl) acrylic acid (0.01 mol) in dry benzene (20 ml), 1,3-diphenyl-2pyrazolin-5-one (0.01 mol) was added and the reaction mixture refluxed for 10 h. The solid that separated on cooling was crystallized from ethanol to give 1.

4,5-Dihydro-6-(4-ethoxyphenyl-4-[1,3-diphenyl-5(4H)-oxo-pyrazolin-4-yl]-3(2H) pyridazinone (2)

4,5-dihydro-3-hydrazino-6-(4-ethoxyphenyl)-4-(1,3-diphenyl-5-hydrazinopyrazolin-4-yl) pyridazine (10), 4,5-dihydro-3-(aryl methylene hydrazonyl)-6-(4-ethoxyphenyl)-4-(1,3-diphenyl-5-arylmethylene hydrazonylpyrazolin-4-yl) pyridazine (12a-d), *General procedure*

To a solution of 1, 7, or 10 (0.01 mol) in ethanol (20 ml), hydrazine hydrate, *o*-chlorobenzaldehyde, *p*-anisaldehyde, *m*-chlorobenzaldehyde, or benzaldehyde (0.01 or 0.02 mol) was added and the reaction mixture refluxed for 5 h. The solid that separated on cooling was crystallized from a suitable solvent to give 2, 10, and 12a-d, respectively.

6-(4-Ethoxylphenyl)-4-[1,3-diphenyl-5(4H)-thioxo-pyrazolin-4-yl]-3(2H) pyridazinethione (4)

A solution of 2 (0.01 mol) and *Lawsson* reagent (0.03 mol) in dry xylene (50 ml) was boiled under reflux for 6 h. The reaction mixture was filtered while hot and then concentrated. The product which separated on cooling was crystallized from ethanol to give **4**.

Compound	Mp (°C)	Yield	Formula ^a	
-	(solvent)	(%)		
1	195	50	$C_{27}H_{24}N_2O_5$	
	(Ethanol)			
2	136	58	$C_{27}H_{24}N_4O_3$	
	(pet. ether 100-120)			
4	234	70	$C_{27}H_{22}N_4OS_2$	
	(Ethanol)			
5	125	72	$C_{41}H_{34}N_4OS_2$	
	(Benzene)			
6	235	69	$C_{27}H_{20}N_4OS_2$	
	(Benzene)			
7	157	44	$\mathrm{C_{27}H_{22}Cl_2N_4O}$	
	(Ethanol)			
8	135	75	$C_{29}H_{28}N_4O_3$	
	(Ethanol)			
9	185	63	$C_{35}H_{38}N_6O_3$	
	(Ethanol)			
10	156	67	$C_{27}H_{28}N_8O$	
	(Benzene)			
12a	266	88	$C_{41}H_{34}Cl_2N_8O$	
	(Acetic acid)			
12b	260	80	$C_{43}H_{40}N_8O_3$	
	(Ethanol)			
12c	275	85	$C_{41}H_{34}Cl_2N_8O$	
	(Acetic acid)			
12d	279	86	$C_{41}H_{36}N_8O$	
	(Ethanol)			
14	156	48	$C_{27}H_{22}N_{10}O$	
	(Ethanol)			

Table 2. Physical data of prepared compounds

^a All compounds gave satisfactory elemental analyses (C, H, N)

6-(4-Ethoxyphenyl)-4-(5-benzylthioxy-1,3-diphenyl-pyrazolin-4-yl)-3-benzylthioxypyridazine (5)

- (a): To a solution of sodium hydroxide (0.02 mol) and phenyl methanethiol (0.02 mol) in water (2.5 ml), cooled in ice, compound 7 (0.01 mol) and dioxan (0.6 ml) were added. The mixture was stirred for 2 h and the separated solid was crystallized from benzene to give 5 (63% yield).
- (b): To a solution of sodium hydrogen carbonate (0.02 mol) and dithione 4 (0.01 mol) in water (10 ml), cooled in ice, benzylchloride (0.02 mol) and acetone (5 ml) were added. After stirring for 2 h at room temperature, the mixture was evaporated under reduced pressure. The separated solid was collected and crystallized from benzene to give 5 (72% yield).

8-(4-Ethoxyphenyl)-1,3-diphenylpyrazolo[3,4-c]dithiino[4,3-e]pyridazine (6)

A solution of iodine (0.02 mol) in 5% aqueous KI solution (100 ml) was added dropwise with stirring to a solution of 4(0.01 mol) in 10% aqueous sodium hydroxide (10 ml) until the color of iodine persisted. The solid formed was filtered off and crystallized from benzene to give 6.

3-Chloro-4,5-dihydro-6-(4-ethoxyphenyl)-4-(5-chloro-1,3-diphenyl-pyrazolin-4-yl) pyridazine (7)

A mixture of 2 (0.01 mol) and POCl₃ (10 ml) was gently refluxed for 30 min, cooled, and treated with crushed ice. The precipitated solid was filtered and crystallized from ethanol to give 7.

4,5-Dihydro-3-methoxy-6-(4-ethoxyphenyl)-4-(5-methoxy-1,3-diphenyl-pyrazolin-4-yl) pyridazine (8)

Compound 7 (0.01 mol) was added to sodium methoxide solution (from sodium (0.02 mol) in absolute methanol (50 ml)). The mixture was refluxed for 2 h and then evaporated under reduced pressure. To the residue, water (10 ml) was added, insoluble matter was filtered off and crystallized from ethanol to give 8.

4,5-Dihydro-3-morpholino-6-(4-ethoxyphenyl)-4-(5-morpholino-1,3-diphenyl-pyrazolin-4-yl) pyridazine (9)

A mixture of 7 (0.01 mol) and morpholine (0.025 mol) in 50 ml of dry toluene was refluxed for 3 h. The amine hydrochloride was filtered off and the organic filtrate was evaporated under reduced pressure. To the residue water (10 ml) was added; insoluble matter was filtered off and crystallized from ethanol to give 9.

4,5-Dihydro-6-(4-ethoxyphenyl)-4-[1,3-diphenyl-triazolo(4,5-b)pyrazolin-4-yl]tetrazolo[2,3-b]pyridazine (14)

- (a): A mixture of 7 (1.0 g), sodium azide (2.0 g), water (5 ml), and N,N-dimethylformamide (20 ml) was boiled for 2 h and cooled. The solid obtained upon dilution with water was filtered and crystallized from ethanol to give 14 (48% yield).
- (b): An aqueous solution of NaNO₂ (0.03 mol in 10 ml H₂O) was added dropwise with stirring to a solution of 10 (0.01 mol) in 4 M acetic acid (10 ml); stirring was continued for 1 h. The solid formed was filtered and crystallized from ethanol to give 14 (45% yield).

References

- [1] Kamiya S, Nakamura A, Itai T, Koshinuma K, Okusa G (1966) Yakugaku Zasshi 86: 1099
- [2] Sayed GH, Ismail AA, Hashem Z (1984) Egypt J Chem 27: 757
- [3] Sayed GH, Ismail AA, Hashem Z (1984) J Chem Soc Pak 6: 95
- [4] Sayed GH, Ismail AA, El-Nagdy S, Mohamed SM (1986) Egypt J Chem 29: 433
- [5] Sayed GH, El-Mobayed M, El-Shekeil AG, Abdel-Ghani E (1990) Indian J Chem 29B: 72
- [6] Sayed GH, El-Mobayed M, El-Shekeil AG, Abdel-Ghani E (1991) Egypt J Chem 34: 73
- [7] Papa D, Schwenk E, Villani F, Klingsberg E (1948) J Am Chem Soc 70: 3356
- [8] Bal'yan Kh V, Shtangeev AL (1954) Zh Obshch Khim 24: 238
- [9] Capparelli MP, Deschepper RE, Swenton JS (1987) J Org Chem 52: 4953

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